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General Chemistry 10th Edition by Darrell Ebbing (Author), Steven D. Gammon (Author) 4.1 out of 5 stars 65 ratings. ISBN-13: 978-1285051376. ... Darrell Ebbing has taught general chemistry for more than thirty years and is now retired from Wayne State University. He received his Ph.D. in physical chemistry from Indiana University.

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the adoption of IUPAC periodic table conventions, along with changes to the discussion of d orbitals.

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MSC: general chemistry 16. Which of the following compounds would be expected

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to have the highest melting point? A) NCl<sub>3</sub> B) OCl<sub>2</sub> C) MgCl<sub>2</sub> D) LiCl E) CCl<sub>4</sub>  
ANS: C PTS: 1 DIF: moderate REF: 9.1  
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The eleventh edition was carefully reviewed with an eye toward strengthening the content available in OWLv2, end-of-chapter questions, and updating the presentation. Nomenclature changes and the adoption of IUPAC periodic table conventions are highlights of the narrative revisions, along with changes to the discussion of d orbitals. In-text examples have been reformatted to facilitate learning ...

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