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explain observations in terms of the laws and theories of chemistry, using procedures that are accepted by the scientific community.

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terms of the laws and theories of chemistry, using procedures that are accepted by the scientific community.

1.1 Chemistry in Context - Chemistry

1.1: Chemistry in
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as “the central
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The Scientific Method.
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Definitions for the book
Chemistry in Context;
Chapter 5; UCF CHM
1020. Terms in this set
(41) Electronegativity.

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is a measure of the attraction of an atom for an electron in a chemical bond. Polar Covalent Bond. a covalent bond in which the electrons are not equally shared but rather are closer to the more electronegative atom.

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ethical issues.

Chemistry in Context by American Chemical Society

(a) When two electrons are removed from the valence shell, the Ca radius loses the outermost energy level and reverts to the lower $n = 3$ level, which is much smaller in radius. (b) The +2 charge on calcium pulls the oxygen much closer compared with

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K, thereby increasing the lattice energy relative to a less charged ion.

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