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the region of high probability that is  
occupied by an individual electron as it  
travels with a wavelike motion in the  
three dimensional space around one of

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two or more associated nuclei. This activity was created by a Quia Web subscriber.

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\_\_\_\_\_ are formed by covalent bonding and are represented by molecular formulas. Molecules: Formula units are formed by \_\_\_\_\_ bonding and are represented by empirical formulas. ionic:

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Nonpolar Covalent Bond: a bond in which 2 like metallic atoms share a pair of electrons equally: Polar Covalent Bond: a bond in which 2 different atoms share ...

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Holt Chemistry Chapter Resource File 6



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(P) [0030414393... Covalent and Metallic Bonds continued PROPERTIES OF MOLECULES Remember that an atom is the smallest piece of an element that still has the properties of that element. In the same way, a molecule is the smallest piece of a covalent compound that has the properties of that compound.

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## **Holt Chemistry Bonds Compounds Concept Review Answers**

Most organic chemicals are covalent compounds, which is why we introduce organic chemistry here. By convention, compounds containing carbonate ions and bicarbonate ions, as well as carbon dioxide and carbon monoxide, are not

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considered part of organic chemistry,  
even though they contain carbon.

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