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vocab and main

concepts. Terms in this

set (68) valence

electron. the electrons

on an atom that can be

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gained or lost in a chemical reaction.
covalent bond. neutral atoms combine to share electrons.

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Chemistry Chapter 6 Test. A bond's ionic character is more than 50% ionic if the electronegativity difference between the two atoms is greater than... A covalent bond in which there is an unequal attraction for the shared electrons is... Atoms with a strong attraction for electrons they share with another atom

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Counting Atoms or

Molecules by Moles The

number of atoms or

molecules we will use

is 6.022×10^{23} and we

call this a mole 1 mole

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atoms potential energy leading to stability and bond formation ; polar nonpolar covalent bonds ; difference in electronegativity ionic character ; using electronegativities, determine if a bond is ionic, polar or nonpolar ...

PPT - Modern Chemistry Chapter 6- Chemical Bonding

...

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6 - Middle School Chemistry

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Terms in this set (36)

Best explains the fact that the change in internal energy equals the difference of the final value and the initial value. Internal energy is a state of function.

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Reaction ...

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Material for test #2 .

Chapter 4 Chapter 6,

part a Chapter 6, part b

Chapter 6, part c

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reactions , Material for

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2 Test 2, key Test 3
Test 3, key

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solution more or less

takes on the

characteristics of the

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is commonly the major fraction of the mixture.

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Choice Identify the

choice that best

completes the

statement or answers

the question. ____ 1. A

mutual electrical

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attraction between the nuclei and valence electrons of different atoms that binds the atoms together is called a(n) a. dipole. c. chemical bond. b. Lewis structure. d. London force. ____ 2.

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Electron Configuration
electrons in the outermost layer of the orbitals ...

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